## A MILD FOUR-CARBON HOMOLOGATION OF 4-FORMYL-AZETIDINONES<sup>1</sup>

Christina Bodurow\* and Michael A. Carr Lilly Research Laboratories, Eli Lilly and Company,Lilly Corporate Center Indianapolis, Indiana 46285

**SUMMARY**: A mild, four-carbon homologation of a 4-formyl-substituted azetidinone provides a route to compound I, a key intermediate in the total synthesis of the carbacephem antibiotics. The route features homologation with phosphorane IV and facile dioxenone opening in the final step of the sequence.

Beta-keto ester I (R"=*p*-nitrobenzyl) was a key intermediate in two recently reported syntheses of carbacephem antibiotics.<sup>2</sup> A route to I different from those previously reported is shown in Scheme I. Racemic azetidinone II can be obtained through a reliable, high-yielding sequence (*vide infra*), but its reactivity towards four-carbon homologating reagents, especially those which **Scheme I** 



require strong base for activation, is not well-documented. Proton abstraction (HA, HB, HC) and beta-lactam ring-opening may be competing pathways for reaction with a strongly basic reagent. Reported herein is a methodology for the successful homologation of II without epimerization of H<sub>A</sub> or H<sub>B</sub>. Hence completion of the sequence to afford the key intermediate I was achieved.

The following beta-keto ester equivalents were synthesized and tested in the coupling reaction with I:



A mixture of products resulting from both alpha and gamma coupling was observed in the reaction with IIIa.<sup>3</sup> The reaction of the dianion of IIIb<sup>4</sup> resulted in degradation of the azetidinone nucleus. The homologating reagent which gave the desired results was dioxenone IV.<sup>5,6</sup> A short, simple synthesis of IV is being reported.<sup>6</sup>

Scheme II describes the synthesis of the 4-formyl azetidinones required for homologation.<sup>7</sup> Scheme II



V=phenoxyacetyl

Both N-phenyl (PMP=*para*-methoxyphenyl) and N-benzyl (DMB=dimethoxybenzyl) protected azetidinones were tested in the sequence. Imines Va and Vb are crystalline solids formed upon condensation of cinnamaldehyde and the appropriate amine (Va: 95% yield; Vb: 74% yield). Staudinger-type [2+2] cycloaddition of imine with the acid chloride of N-phthaloylglycine (triethylamine,  $0^{\circ}$  C)<sup>8</sup> gave azetidinones VI (VIa, 80%; VIb, 43%). Removal of the phthaloyl

group<sup>9</sup> from IV and reprotection of the 3-amino functionality with phenoxyacetyl chloride under Schotten-Baumann conditions afforded crystalline intermediates VIIa (74%) and VIIb (79%). The 4-cinnamyl side-chain was converted via ozonolysis to the requisite 4-formyl group (R=PMP, 73 %; R=DMB, 86%).

Scheme III details the conversion of the key 4-formyl azetidinones VIII to form the penultimate intermediate I. In the critical homologation reaction, the solids VIII and IV were combined in Scheme III



acetonitrile, stirred at ambient temperature for 16h, the solvent was removed under reduced pressure and the product IX was crystallized from the resulting residue (R= PMP, 76%; R=DMB, 49%). Compounds IXa and b were hydrogenated (for IXa, 5% Pd/C, EtOAc, 18 h, 66%; for IXb, 10% Pd/C, 5 d, EtOAc, 88%). The N-protecting group was oxidatively removed using ceric ammonium nitrate<sup>10</sup> in the case of Xa (67%) and buffered potassium persulfate<sup>11</sup> in the case of Xb (13%). At this point both sequences had converged upon the common intermediate XI. Samples of both oxidation reaction products were identical to each other in all respects.

By simply combining XI with an equivalent of p-nitrobenzyl alcohol in toluene, warming to 110<sup>o</sup> C, cooling and filtering, the crystalline racemic product I was obtained in 57% yield. The material thus obtained was analytically pure and identical in all respects (except for optical activity) with authentic samples of I which had been generated by the previously reported route.<sup>2b</sup> The synthesis detailed in this paper could be made asymmetric by employing a chiral auxillary in the cycloaddition step (as in footnotes 2a,b), or by incorporating a resolution into the conversion of VI to VII.

The use of phosphorane IV provides an efficient means of incorporating the beta-keto ester functionality into base-sensitive azetidinones such as VIII. We have thus demonstrated that this

methodology provides a workable alternative for the construction of precursors to the carbacephem antibiotics.

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## **References and Notes**

- Bodurow, C. C., Carr, M. A. Abstracts of Papers, 196th National Meeting of the American Chemical Society, Los Angeles, CA; American Chemical Society: Washington, DC, 1988; ORGN 0075.
- (a) Evans, D. A.; Sjogren E. B. *Tetrahedron Lett.* **1985**, *26*, 3787; (b) Bodurow C. C.; *et. al. Abstracts of Papers*, 196th National Meeting of the American Chemical Society, Los Angeles, Sept. 25-30, 1988; Washington, D. C., 1988; ORGN 0076.
- 3. Pietrusiewicz, K. M.; Monkeiwicz, J. Tetrahedron Lett. 1986, 27, 739.
- 4. Bolaski, P.; Pietrusiewicz, K. M.; Monkiewicz, J.; Koskuk, J. Tetrahedron Lett. 1980, 21, 2287.
- 5. Boeckman, Jr., R. K.; Thomas, A. J. J. Org. Chem. 1982 47, 2823.
- 6. Bodurow, C. C.; Moore, L.C.; Carr, M. A. Organic Prep. and Proc. Int., 1989, 0000.
- All compounds reported in this Letter were fully characterized and their structure assignments supported by <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, IR, MS and elemental analysis.
- For a review, see Holden, K. G. in "Chemistry and Biology of Beta-Lactam Antibiotics;" Morin, R. B. and Gorman, M., Eds; Academic Press: New York, 1982; Vol. 2, Chapter 2.
- 9. Kukolja, S.; Lammert, S. R.; Ellis, A. I. Croatica Chem. Acta 1977, 49, 779.
- For leading references, see (a) MacCauley, D. D.; Trahanovsky, W. S. J. Am. Chem. Soc. 1973,38, 1497; (b) Jacob, P. III; Callery, P.S.; Shulgin, A. T.; Castagnoli, N., Jr. J. Org. Chem. 1976, 41, 3627 (c) The conditions in this paper closely followed: Hart, D. J. J. Antibiotics, 1987, 309.
- For leading references, see (a) Needles, H. L.; Whitfield, R. E. J. Org. Chem. 1964, 29, 3632; (b) Huffman, W. F.; Holden, K. G.; Buckley, T. F. III; Gleason, J. G.; Su, L. J. Am Chem. Soc. 1977, 99, 2352; (c) The conditions in this paper closely followed: Kametani, T.; Nakayama, A.; Matusumoto, H. Chem. Pharm. Bull. 1983, 31, 2578.

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